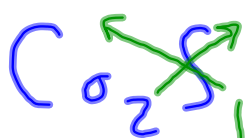
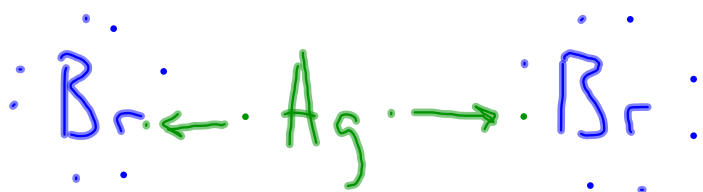
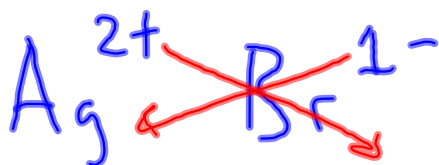


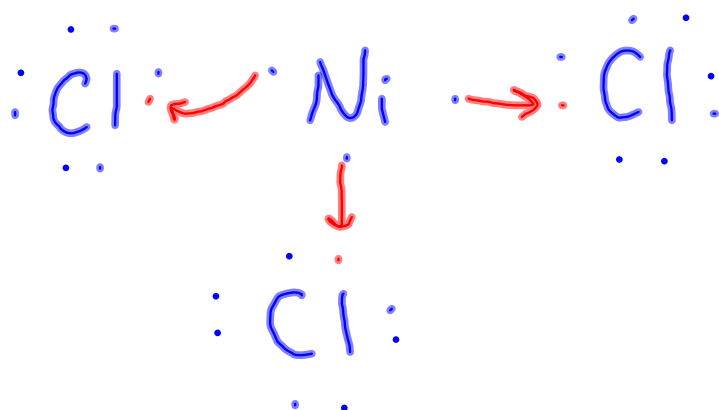
Homework Review:

- $\text{Co}_2\text{S} \rightarrow \text{non-metal}$
 $\hookrightarrow \text{metal}$

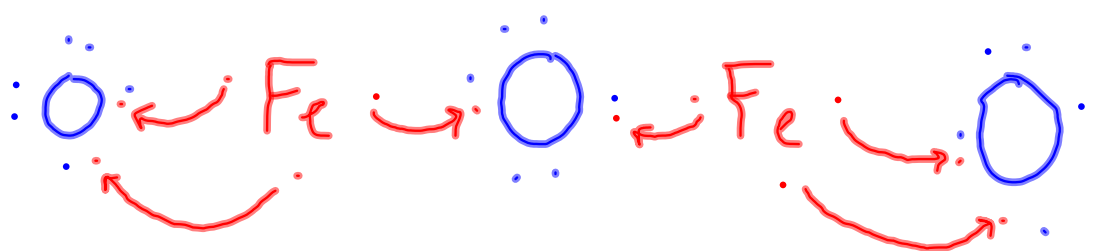
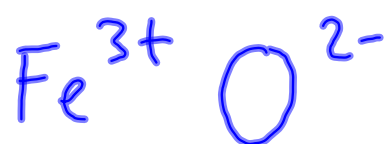


- Silver (II) Bromide





• Iron (III) Oxide



• Metallic Bonding:

- Bonding between metals
- All metals want to give away electrons (positive oxidation numbers)
- Don't transfer electrons specifically; rather electrons move freely between atoms → this is why metals are so good at transferring heat and electricity